A XANES study of chromophores: the case of black glass

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This report summarizes the results published in a paper recently submitted to Analytical Chemistry [1]. At DORIS Beamline L we examined several Roman black glass fragments using Fe K-edge X-ray absorption near edge (XANES). The selected archaeological glass samples cover the period 1st-5th century AD along ten different sites of the North Western provinces of the Roman Empire. The main goal was to determine the Fe³⁺/ΣFe, since this ratio influences glass-colour and can result in hues ranging from brown (in presence of sulphur) over green to blue or optionally even colourless glass. 27 Fragments from vessel glass and glass jewelry were selected from different periods in Roman history in which black glass was considered fashionable.

At DORIS Beamline L a Si(111) double crystal monochromator was used, having an energy resolution ΔE/E of about 10⁻⁴. A transmission XANES spectrum was recorded from a metallic Fe reference foil and used to provide an accurate energy calibration for all spectra; first inflection point of the Fe-K edge was set to 7112 eV. A slight detuning of the two monochromator crystals was used to remove high energy harmonics from the incident X-ray beam. FeCl₃ and FeSO₄ reference compounds were recorded in transmission mode and provided reference spectra for ferrous (II) and ferric (III) iron. Self-absorption was limited by diluting these powdered compounds to app. 2 wt % with cellulose and pressing it into a pellet. In addition two glasses were recorded in fluorescence mode with a different total amount of Fe (0.1 wt% and 0.6 wt%) but an identical Fe³⁺/ΣFe ratio (0.67). The fluorescence yield was recorded as a function of the incoming X-ray energy using a Vortex detector at an angle of 45º with respect to the incoming beam, while also the sample surface is oriented at 45º to the incoming beam. ROI fitting was performed around the Fe Kα line (6113-6692 eV). XANES spectra were collected form ~32 eV below to ~130 eV above the Fe-K edge using different energy intervals (7090 eV - 7105 eV: 0.25 eV, 7105 eV - 7120 eV: 0.1 eV, 7120 eV - 7250 eV: 0.25 eV); a measuring time of 1 s was used for each energy step resulting in a ~20 min measuring time per XANES spectrum. Three repeats were recorded for each Fe model compound in order to acquire virtually noise-free spectra. Measurements on real glass samples were performed under the same conditions, however the focused primary beam was made as large as possible, circa 780 x 520 µm². For all XANES spectra, the normalisation was performed by means of the software package ATHENA. An edge-step normalisation was performed by a linear pre-edge subtraction and by regression of an (in general) third degree polynomial beyond the edge [2].

The pre-edge peak is extracted using an arctangent function to describe the background (Fig. 1). In literature different functions have been employed for the description of the pre-edge features [3]. When the same mathematical model is used throughout the entire data set, the conclusions will be consistent. Problems may arise, however, when comparing the data to literature values [4]. For the glass samples, the Fe-K pre-edge features have been fitted using two Voigt peaks with 2 eV of average width [3-7]. For the reference compounds, it was necessary to include a third peak function to obtain good fits. All peaks were constrained to have a 50% Lorentzian-50% Gaussian shape.
The centroid of the extracted pre-edge for an certain oxidation state always occurs at the same position irrespective to the coordination. Therefore its position can be used to determine the $\text{Fe}^{3+}/\Sigma\text{Fe}$, which is a relevant parameter testifying to the redox conditions during the production of coloured glass.

The results obtained suggest that in an earlier stage (first century AD) black glass was produced in a single step, under controlled reducing conditions ($\text{Fe}^{3+}/\Sigma\text{Fe} \approx 0.17$). After the second half of the 2nd century AD, instead, the $\text{Fe}^{3+}/\Sigma\text{Fe}$ values (0.4-0.5) are similar to what Arletti et al. [8] found for naturally coloured glass. This corroborates the hypothesis that large quantities of iron were added in a secondary step to turn naturally coloured glass into black. This was most likely achieved without requiring specific control of the redox conditions in the furnace.

References